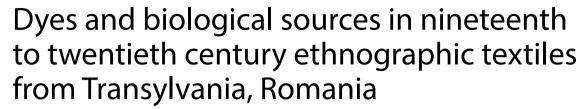


# **RESEARCH ARTICLE**

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Irina Petroviciu<sup>1\*</sup>, Iulia Teodorescu<sup>2</sup>, Florin Albu<sup>3</sup>, Marian Virgolici<sup>4</sup>, Eugenia Nagoda<sup>5</sup> and Andrei Medvedovici<sup>6</sup>

## **Abstract**

Liquid chromatography with UV-Vis and mass spectrometric detection (LC-DAD-MS) was applied to the identification of dyes and biological sources in samples from nineteenth to twentieth century ethnographic textiles from ASTRA National Museum Complex, Sibiu, Transylvania. The objects are part of the Romanian traditional costume and are among the first to be acquired for the museum collections, around 1905. Oral and written information mention such objects as homemade, with nearby materials, while literature mentions a significant number of local vegetal sources as being used for textile dyeing. The analytical protocol developed, based on the combined use of the UV-Vis and mass spectrometric detectors to associate the information and distinguish between major and minor dyes, facilitates a clear attribution of the dyes and biological source/sources used. Other techniques, such as X-ray spectroscopy and FTIR-ATR were successfully used to identify inorganic dyes, which may not be detected by LC-DAD-MS, as was the case of Prussian blue. A large number of biological sources was identified in the studied objects, both local and imported. The local sources identified include dyer's broom (Genista tinctoria L.), sawwort (Serratula tinctoria L.), young fustic (Cotinus coggygria Scop.), Rhamnus berries, emodin based dyes (Rhamnus, Rheum, Rumex sp.) and woad (Isatis tinctoria L.), in perfect correlation with literature which states that local dyes were still in use in the period under discussion. Carminic acid containing insects (Dactylopius coccus Costa and Porphyrophora sp.) and redwood type Caesalpinia species should be considered a result of trade. Almost all the natural and synthetic dyes detected are frequently mentioned in a collection of recipes published by the Romanian Academy, in 1914. The richness in colours in belts, the use of insect dyes in shirts decoration and the large amount of cotton in shirts are illustrative for the owners' status. The study provides a better valorisation of the Romanian traditional costume as witness of the rural society at the end of the nineteenth to beginning of the twentieth century and emphasizes the usefulness of chemistry in cultural heritage dedicated applications.

**Keywords:** Natural dyes, Liquid chromatography, Mass spectrometry, Ethnographical textiles, Belts, Shirts, Decoration, Romania

Full list of author information is available at the end of the article



<sup>\*</sup>Correspondence: petroviciu@yahoo.com

<sup>&</sup>lt;sup>1</sup> National Museum of Romanian History (MNIR), Calea Victoriei 12, 030026 Bucharest. Romania

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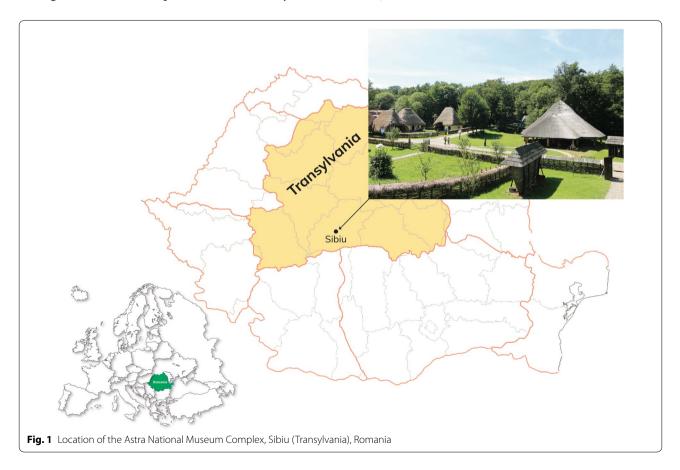
#### Introduction

Ethnographical textiles, from which the oldest date back to the second half of the nineteenth century, represent an important part of the Romanian heritage. As witnesses of the rural life in this period, ethnographical textiles have been studied mainly considering artistic, historic and anthropological criteria, with only rare references to the materials used [1-4]. It was only in the last 20 years that more particular studies were dedicated to fibers, techniques and to the biological sources used for dyeing [5–8]. According to the results obtained in the first study dedicated to the identification of natural dyes in ethnographical textiles from Romanian collections, in 1997 [7], local biological sources, imported natural dyes and synthetic dyes co-existed in the mentioned period. The results of the above mentioned work—performed within a joint research between Romanian institutions and the Royal Institute for Cultural Heritage in Brussels (KIK/ IRPA)—is in perfect agreement with literature [9].

In the present study, dye analysis on selected textiles from the same period, preserved in the ASTRA National Museum Complex, Sibiu, Transylvania (Fig. 1) are discussed. The objects (blouses and belts) are part of the Romanian traditional costume and are among the first to be acquired for The Transylvanian

Association for Romanian Literature and the Culture of the Romanian People (ASTRA) collections, around 1905 [10]. ASTRA was a cultural association founded in 1861 in Sibiu, which had an important role in the cultural life and the movement of national awakening for the Romanians in Transylvania, its museum being the predecessor of ASTRA National Museum Complex [10]. Oral and written sources mention such objects as homemade, with nearby materials (wool, hemp, local dye sources). Literature mentions wool and hemp as being used since Neolithic and late Neolithic respectively [5] while silk as being produced in South East Romania, Dobrogea region, since 1384 [11]. Apart from the esthetic value, the colorful handmade embroidery made by natural dyed silk and wool reveal a persons' statute while the use of cotton would indicate dating after 1900 [12]. The nineteenth century belts are worked in a unique technique of Sprang [11].

For late nineteenth century objects, dyes identification is extremely important, if we consider that natural dyes were used since antiquity and mauveine (or mauve), the first synthetic dye was discovered in 1856 [13]. Moreover, detection of specific combinations of natural dyes allows attribution to the most probable biological sources [14, 15]. If we take into account that



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local sources were used before dyes became available through commerce, and that many trade routes are now documented, as being connected to historical events and geographical discoveries, identification of a specific source would be helpful to understand more about the context an object was created.

For several years since its development by Wouters, in 1985 [16], liquid chromatography with UV-Vis (diode array) detection was the standard method for dye analysis [17-20]. More recently, liquid chromatography with UV-Vis and mass spectrometric (MS) detectors became the ideal configuration for natural dyes investigation [21-24]. The use of mass spectrometers brings a new criteria-the molecular ion (for simple MS detectors) and, for MS/MS systems, the product ion scan. Mass spectrometers also allow lower detection limits for most of the dyes used in historical and archaeological textiles, due to the increased sensitivity achieved by selectivity (more often through decreasing the noise level, as a direct result of the induced selectivity) [25]. The MS/MS configurations were also proved to be useful to discover the molecular structure of unknown dyes [26]. In the last years, ultra high pressure liquid chromatography (UHPLC) uses smaller particle size columns to achieve increasingly accurate chromatographic results [27, 28].

In the present study, dye analysis were performed by liquid chromatography with UV–Vis and mass spectrometric detection (LC–DAD–MS), according to a procedure described in detail in earlier publications [29, 30]. The method was applied on a selection of samples from belts and shirts, from the collection of ASTRA Museum Complex, Sibiu, Transylvania.

## **Materials and methods**

#### Samples and sample preparation

Samples about 0.5 cm long ( $\sim$  3 mg) were taken during the conservation procedures from 7 textiles (3 belts and 4 shirts), dated nineteenth to twentieth century, belonging to ASTRA Museum, Sibiu, Romania (Fig. 1). A total of 31 samples were available, 12 from belts and 19 from shirts decoration (Fig. 2). Fibres were first observed under the stereo microscope, at  $10-80\times$  and then non-destructively analysed by attenuated total reflectance infrared spectroscopy (FTIR-ATR). In one case only, when the presence of Prussian blue was suggested by FTIR-ATR, elemental analysis of the coloured fibre was determined by X ray fluorescence spectrometry (XRF).

Dyes extraction was made by acid hydrolysis, according to the method developed by Wouters [16]. Although limited, as it decomposes glycosides to their parent aglycons in the case of yellow flavonoid dyes [31, 32], this method was preferred because a dedicated database was

already in use. 250  $\mu$ L mixture 37% HCl/CH<sub>3</sub>OH/H<sub>2</sub>O 2:1:1 (v/v/v) were added on each fibre and the mixtures were kept at 100 °C for 10 min. The solutions were evaporated to dryness in a vacuum desiccator. Each sample was redissolved in 100  $\mu$ L solution CH<sub>3</sub>OH/H<sub>2</sub>O 1:1 (v/v) and centrifuged at 12,000 rpm for 10 min. The supernatants were transferred in chromatographical vials and injected into the chromatographical system. For the visual blue and green samples, a second extraction in 100  $\mu$ L dimethyl sulfoxide (DMSO) was made, and the samples were kept at 80 °C for 10 min. The two solutions were analysed together. More details on sample preparation and the instrumentation used were described in earlier publications [29, 30].

#### Fibre investigation

A Nikon SMZ 1000 stereomicroscope was used for fibres observation. For documentation and images collection, the microscope was coupled with a Nikon DSLR camera, model D3100 Kit AF-s 18–55 mm VR DX. Further investigation on the fibres was made by infrared spectroscopy (FTIR-ATR) by using a Bruker Optics Alpha spectrometer equipped with a Platinum ATR single reflection diamond ATR module. Spectra were acquired in the 4000–400 cm<sup>-1</sup> domain, with a resolution of 4 cm<sup>-1</sup>. Spectra collection and data processing were made with a dedicated software, Opus 7.0.

## Dye analysis Database

Dyes were identified according to retention, UV–Vis and mass spectrometric data, as compared to information collected on standards. Analysis of standard dyed fibres (fibres dyed in the laboratory with known biological sources, by following traditional dyeing methods) and information available in literature are the basis of biological sources attribution [14, 15]. The biological sources of dyes are given in Table 1, while retention, UV–Vis and mass spectrometric data of the dyes detected, are described in Table 2.

#### Instrumentation

Samples were analysed by liquid chromatography with diode array and (triple quadrupole) mass spectrometric detection (LC-DAD-MS). An Agilent 1260 LC system was used, composed of the following modules: quaternary pump (Model G1311C), automatic injector (G1367E) and column thermostat (G1316C). The diode array detector (G4212A) and the triple quadrupole mass spectrometer (G6410B) were serially connected. The latter was using an ESI ionization source (ESI, Model G1948B), operated under negative ion monitoring mode.

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**Fig. 2** Examples of objects (belts and shirts with details on decorations) from which 31 dyed fibre samples were withdrawn. Left: Belt 554P, ensemble and detail (the belt is 254 cm long and 40 cm wide, the sphere decoration is 4 cm in diameter). Right: Shirt 20P, ensemble and details (the shirt is 39 cm long and 75 cm wide, the sleeve is 37 cm wide, neck diameter 14 cm)

#### Chromatographic separation

A Zorbax C18 column, 150 mm L  $\times$  4.6 i.d.  $\times$  5  $\mu m$  d.p. was used, thermostated at 40 °C. The mobile phase consisted in a mixture of aqueous 0.2% (v/v) formic acid (solvent A) and methanol/acetonitrile (1:1, v/v, solvent B). Gradient elution was applied according to the following profile: at 0 min, 15% solvent B; from 0 to 5 min, linear increase to 25% solvent B; from 5 to 10 min, linear increase to 55% solvent B; from 10 to 16, linear increase to 100% solvent B; from min 16 to 18, constant at 100% solvent B; and step jump at 15% solvent B, with a 4 min re-equilibration step (period between runs). The flow rate was set at 0.8 mL/min. 5  $\mu$ L were injected for each sample, from the 100  $\mu$ L volume resulting from sample preparation.

#### Detection

UV–Vis spectra were acquired with a DAD detector which was placed between the column and the MS ion source. Spectra were collected over the 190–640 nm range, with a resolution of 2 nm. A triple quadrupole MS detector was used, which was operated in negative ion monitoring mode, with the following ESI operation

parameters: drying gas temperature 350 °C; drying gas flow 8 L/min; pressure of the nebulising gas 40 psi; Vcap 2500 (—). The triple quadrupole used MS2 type Scan when used as a single MS instrument; the data storage was set on profile and the peak width at 0.07; fragmentor 135 V;  $\Delta$ EMV 400 V; The scanning interval for the mass to charge ratio (m/z) was between 100 and 600 a.m.u., acceleration voltage on the collision cell: 7 V; Dwell Time 500 ms. When working in the product ion scan mode, the following parameters were used: start mass 50 a.m.u.; end mass 600 a.m.u.

### Data processing

Agilent MassHunter Quantitative Analysis B.06.00 software was used to control the chromatographic system, for data acquisition and processing. The analytical procedure was described in detail in an earlier publication, where an ion trap mass spectrometer was used instead of the triple quadrupole [29]. Each sample was first analysed with single MS detection exploited in the Full Scan mode and the resulted data was processed by extracting chromatograms, according to the molecular ions of the dyes in the database. If necessary, a second

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Table 1 The biological sources discussed in the present study, the corresponding dyes and their abbreviation

Biological source		Dye component	Abbreviation	
Common name	Latin name(s)			
Berries	Rhamnus sp. (berries)	Quercetin	qu	
		Kaempferol	kpf	
Carminic acid based insects	Dactylopius coccus	Carminic acid	ca	
	Porphyrophora hameli	Flavokermesic acid (C glycoside)	dcll	
	Porphyrophora polonica	Flavokermesic acid	fk	
		Kermesic acid	ka	
Dyer's broom	Genista tinctoria L.	Luteolin	lu	
		Genistein	ge	
		Apigenin	ар	
		Chrysoeriol	chry	
		Diosmetin	dios	
Emodin based dye	Rhamnus, Rheum, Rumex sp.	Emodin	em	
Indigoid dyes	Indigofera sp./Isatis tinctoria	Indigotin	ind	
		Indirubin	inr	
Redwood type	Caesalpinia sp.	Brazilein	bra	
		Soluble redwood (urolithin C)	srw	
Sawwort	Serratula tinctoria L.	Luteolin	lu	
		3-O-methylquercetin	3-O-methylqu	
		Apigenin	ар	
		Quercetin	qu	
Tannin source	Quercus sp. et al.	Ellagic acid	ea	
Young fustic	Cotinus coggygria Scop.	Fisetin	fi	
		Sulphuretin	sul	
Weld	Reseda luteola L.	Luteolin	lu	
		Apigenin	ар	
		Chrysoeriol	chry	

injection from the sample was made, by using MS/MS detection. In such a case, the first mass analyzer filters the m/z of the molecular (or major) ion of compounds according to database, while the second mass analyzer is exploited in the Full Scan mode. In most cases, retention, UV–Vis and mass spectrometric data were used for the major compounds identification. The minor ones, associated in the biological source with the major dyes detected, were only recognized based on retention and mass spectrometric data.

#### X ray fluorescence spectrometry (XRF)

A Bruker manufactured S1 TITAN (Model 600) portable XRF spectrometer was used for elemental analysis of sample 20\_P3, for which FTIR-ATR suggested the presence of Prussian blue. The system had the following specifications: 4 W rhodium (Rh) tube (15–50 kV, 5–100  $\mu$ A), silicon drift chamber detector (SDD), 5 mm spot size. The system was used air-path, elemental range Z>12 (Mg).

#### **Results and discussion**

#### Blue dyes

#### **Belts**

Indigotin was detected in all the three blue samples (3551\_P1, 555\_P3, 554\_P1) from the main material in the belts, identified as hemp (Table 3). The detection of indigotin was based on the UV-Vis data at 255 nm (Table 2), correlated with retention. It was also confirmed by the presence of its molecular ion (m/z = 261)[M-H]<sup>-</sup>) in the chromatogram collected with the mass spectrometer in the Full Scan mode, followed by data processing through Ion Extracted Chromatogram (IEC). As minor compound, the presence of indirubin was only established by the MS detection, as suggested the presence of the molecular ion, m/z = 261 a.m.u., in the FS-IEC, in correlation with retention. Identification of indigotin and indirubin put forward the use of an indigoid dye. Several plants containing indigotin precursors are mentioned in literature [14], from which Indigofera species and Isatis tinctoria (woad) are the most well-known. As the later was cultivated in Europe

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Table 2 Retention, UV-Vis and MS data for the dyes detected in the present study

Dye component (abbreviation)	Retention	UV–Vis <sup>a</sup>	[M-H] <sup>-</sup>		
ap	13.7	210; 268; 336	269		
ca	8.8	226; 276; 310; 494	491		
chry	13.9	_	299		
dcII	8.3	288; 434	475		
dios	14.0	-	299		
ea	9.8	254; 366	301		
em	17.5	222; 254; 266; 288; 438	269		
fi	11.3	206; 248; 320; 360	285		
fk	13.5	=	313		
ge	13.4	208; 260	269		
ind	16.2	238; 285; 330; 610	261		
inr	17.0	=	261		
ka	13.4	=	329		
kpf	13.7	204; 266; 366	285		
lu	12.7	208; 254; 266; 348	285		
3-O-methylqu	12.9	=	315		
qu	12.7	208; 254; 265; 248	301		
srw	10.5	258; 306; 336	243		
sul	12.2	256; 369	269		
Indigo carmine	11.2	250; 288; 340; 615	_		
Prussian blue	Identified by FTIR/ATR $^{\rm b}$ (signal at 2074 cm $^{-1}$ ) and confirmed by the presence of Fe in XRF $^{\rm c}$				

Dyes are listed in alphabetical order. At the end, data on the synthetic dyes detected is given. See Table 1 for the natural dyes biological sources and the dyes names

for a long time and also mentioned in the Romanian literature at the end of the nineteenth century as used for dyeing under the name "drobsor", *Isatis tinctoria* (woad) should be considered as the most possible indigoid dye.

For the two visual blue silk samples from the belts tassel decorations available for analysis (3551\_P3 and 554\_P3), indigo carmine was detected based on the UV–Vis data at 255 nm, correlated with retention. Indigo carmine was also detected in a green silk sample (3551\_P2), together with natural sources of yellow (which will be discussed in the "Yellow dyes" section). Indigo carmine (Acid Blue 74), a semi-synthetic dye derived from indigo by sulfonation (which makes the compound soluble in water) was first discovered in 1740 [33]. It was identified in several textile objects from the last quarter of the eighteenth century to the beginning of the twentieth century [33].

For the other two visual green silk samples no sources of blue were detected.

#### Shirts

Blue organic dyes were detected in 3 visual blue samples available for analysis from the shirts decoration, while in a black sample, Prussian blue, a dark blue pigment, was identified.

Indigotin and indirubin were detected together in one wool sample (44\_P3) which indicates that an indigoid dye was used for dyeing. Indigotin alone was present as major dye in two more samples (20\_P4 and 20\_P5), on silk, from another shirt. In one of them, which has a bluegreen hue, luteolin was also detected, as minor compound. As stated above, *Isatis tinctoria* (woad) should be considered the most probable dyeing source used for all these samples.

In a visual black sample (20\_P3) carminic acid, known as source of red colour, was the only dye present in the UV–Vis and MS chromatograms. Although a search for dyes in the database was made by extracting the chromatograms according to molecular (or major) ions, no other dyes were present in the black sample 20\_P3, neither as major nor as minor component. The database contains most of the dyes mentioned in literature as being used in Europe and Minor Asia. However, the presence of Prussian blue,  $Fe_4[Fe(CN)_6]_3$  was indicated by the typical carbon–nitrogen bond, at 2074 cm<sup>-1</sup> in the attenuated total reflectance infrared spectroscopy (FTIR-ATR) and confirmed by the detection of iron in the elemental analysis by X-ray fluorescence spectrometry (XRF) (Fig. 3).

<sup>&</sup>lt;sup>a</sup> UV–Vis data is given only for the major dyes (dyes identified by DAD in the present study)

b,c See text, "Results and discussion" section/Blue dyes/Shirts

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Table 3 Information about the objects, samples and the results obtained by dye analysis

No.	Object	Sample code	Colour	Fibre	Sample location	Dyes detected	Result
	Belt	3551_P1	Blue	Hemp	Bottom	<u>ind</u> , inr	Indigoid dye
	3551 P (ASTRA)	3551_P2	Green	<u>Silk</u>	Tassel decoration	qu, kpf, indigo carmine	Berries from <i>Rhamnus</i> sp. and indigo carmine
		3551_P3	Blue	Silk	Tassel decoration	indigo carmine	Indigo carmine
		3551_P4	Yellow	<u>Silk</u>	Tassel decoration	<u>lu</u> , 3- <i>O</i> -methylqu, ap, qu	Sawwort
2.	Belt	555_P1	Green	<u>Silk</u>	Tassel decoration	<u>lu, ap</u> , chry	Weld
	555P (ASTRA)	555_P2	Yellow-orange	<u>Silk</u>	Tassel decoration	<u>srw, em, rht</u> , qu	Emodin based dye and redwood type
		555_P3	Blue	Hemp	Interior	<u>ind</u> , inr	Indigoid dye
3.	Belt	554_P1	Blue	Hemp	Interior	ind, inr	Indigoid dye
	554P (ASTRA)	554_P2	Green	Silk	Tassel decoration	<u>lu, ge, ap,</u> 3- <i>O</i> -methylqu, chry, dios	Dyer's broom and sawwort
		554_P3	Blue	Silk	Tassel decoration	<u>indigo carmine</u>	Indigo carmine
		554_P4	Light red	Silk	Tassel decoration	<u>ca, dcll</u> , ka, fk	Carminic acid based insect (possible <i>Dactylopius coccus</i> )
		554_P5	Yellow	Silk	Tassel decoration	<u>fi, sul</u>	Young fustic
4.	Shirt	27_P1	Brown	Silk		<u>ea</u>	Tannin source
	27P Bran (ASTRA)	27_P2	Dark brown	Silk	Shoulder decoration	<u>ea</u>	Tannin source
	(//3/17/)	27_P3	Violet	Silk	Shoulder decoration	<u>ca</u> , dcII, ka, fk, <u>ea</u>	Carminic acid based insect (possible <i>Porphyrophora hamelii</i> ) and a tannin source
5.	Shirt	165_ P2	Brown	Silk	Back, bottom	<u>ea</u>	Tannin source
	165P Rasinari (ASTRA)	165_P3	Violet	Silk	Back, bottom	<u>ca</u> , dcII, ka, fk, <u>ea</u>	Carminic acid based insect (possible <i>Porphyrophora hamelii</i> ) and a tannin source
		165_ P4	Light brown	Silk	Tassel	<u>ea</u>	Tannin source
5.	Shirt	44_P1	Brown	Silk	Neck decoration	ea	Tannin source
	44P Rasinari (ASTRA)	44_P2			Shoulder decoration	<u>ca</u> , dcII, ka, fk, ea	Carminic acid based insect (possible <i>Porphyrophora hamelii</i> ) and a tannin source
		44_P3	Blue	Wool	Neck decoration	<u>ind</u> , inr	Indigoid dye
		44_P5	Light green	Wool	Shoulder decoration	<u>lu</u>	Luteolin based dye
		44_P6	Red	Wool	Neck decoration	<u>ca, dcll</u> , ka, fk	Carminic acid based insect (possible <i>Dactylopius coccus</i> )
		44_P8	Yellow	Silk	Sleeve decoration	<u>lu</u> , ap, chry	Weld (Reseda luteola L.)
7.	Shirt 20P (ASTRA)	20_P1	Pink	Silk	Neck decoration	<u>ca</u> , dcII, ka, fk, ea	Carminic acid based insect (possible <i>Porphyrophora hamelii</i> ) and a tannin source
		20_P2	Light pink	Silk	Neck decoration	<u>ca</u> , ea	Carminic acid based insect and a tannin source
		20_P3	Black	Silk	Shoulder decoration	Prussian blue <sup>a</sup> , ca	Prussian blue and carminic acid based insect
		20_P4	Blue-green	Silk	Shoulder decoration	ind, lu	Indigoid dye and a luteolin based dye
		20_P5	Blue	Silk	Neck decoration	<u>ind</u>	Indigoid dye
		20_P6	Brown	Wool	Neck decoration	=	Natural dyed wool
		20_P7	Ochre yellow	Silk	Neck decoration	<u>lu</u> , ge, ap, 3- <i>O</i> -methylqu chry, dios	Dyer's broom and sawwort

Major dyes (dyes identified by both UV-Vis and MS detectors) are underlined

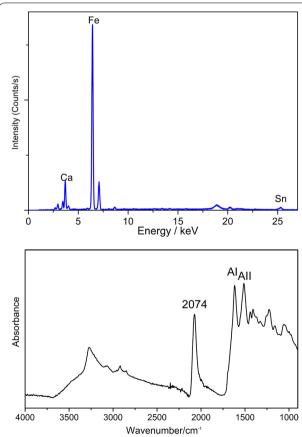
 $<sup>^{\</sup>rm a}~$  Prussian blue was identified by FTIR/ATR (signal at 2074 cm $^{\rm -1}$ ) and confirmed by the presence of Fe in XRF

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## Yellow dyes Belts

A large variety of flavonoid dye sources were detected in the three visual yellow (including one with an orange hue) and three visual green samples from the belts decoration.

The presence of luteolin as major dye was attested in three samples (3551\_P4, 555\_P1, 554\_P2) based on retention, UV–VIS spectra and MS (FS-IEC) spectra (see Table 2 for these values). Luteolin, together with apigenin, are present in many plants, from which three played an important role in textile dyeing: weld (*Reseda luteola* L.), sawwort (*Serratula tinctoria* L.) and dyer's broom (*Genista tinctoria* L.). In order to clearly distinguish between them, identification of other dyes is necessary: chrysoeriol (in the absence of genistein) for weld, 3-O-methylquercetin for sawwort and genistein for dyer's broom. Recent studies also demonstrated that, for the latter, both chrysoeriol and diosmetin should be expected [27]. In the present group of samples, identification of apigenin as major dye and chrysoeriol as minor



**Fig. 3** ATR-FTIR spectrum illustrating the presence of Prussian blue  $(Fe_4[Fe(CN)_6]_3)$ , through the presence of the typical vibration of the C–N bond, at 2074 cm<sup>-1</sup>. The main protein bands, amide A  $(A_A)$  (3269 cm<sup>-1</sup>), amide B  $(A_B)$  (3066 cm<sup>-1</sup>), amide I  $(A_{IJ})$  (1512 cm<sup>-1</sup>) are also evidenced

one, in one sample (555 P1), certifies the use of weld (Reseda luteola L.). As in the case of other minor dyes, chrysoeriol was only detected by MS, FS-IEC according to its molecular ion, m/z = 299 a.m.u. In one other case (3551 P4), apigenin as well as 3-O-methylquercetin were identified as minor dyes which suggests the use of sawwort (Serratula tinctoria L.). Detection of quercetin as minor dye comes to reinforce this supposition [34, 35]. The presence of the three dyes was put forward only by the MS detection, based on the presence of the molecular ions m/z = 269 a.m.u., m/z = 315 a.m.u. and m/z = 301 a.m.u., in the FS-IECs. According to literature [35], the acid hydrolysed extract of sawwort contains the flavones luteolin and apigenin and flavonols quercetin and kaempferol, the two later being susceptible to photo-degradation. 3-O-methylquercetin should be thus considered a marker compound, present in sawwort aged samples, together with luteolin and apigenin. For the other sample where luteolin was detected (554 P2), genistein and apigenin were present as major dyes (retention, UV-VIS at 255 nm and MS), together with luteolin, while 3-O-methylquercetin, chrysoeriol and diosmetin, were minor components (not present in the UV-VIS chromatogram). The presence of the 6 dyes (3 major and 3 minor) suggest that a combination of dyer's broom (Genista tinctoria L.) and sawwort (Serratula tinctoria L.) was used for dyeing. This statement is reinforced by the ratio between genistein and apigenin, two dyes with the same molecular weight (Mw=270, m/z=269 a.m.u. in negative ion mode),which makes them comparable in the IECs. Analysis, performed on several samples where dyer's broom was identified as a unique dyeing source, showed that genistein is present in a larger amount as compared to apigenin, while in the present sample it is the opposite. This could be explained by the extra amount of apigenin, coming from sawwort (Fig. 4). Although the last two samples discussed (3551\_P4 and 554\_P2) have a visual green colour, no source of blue dyes was detected which would indicate that more probable metal mordants were used to give the fibre a green hue.

Quercetin and kaempferol were detected, as major dyes, together with indigo carmine, in one green sample (3551\_P2, also mentioned above as containing blue dyes). The presence of the two flavonoid dyes was established by both UV–VIS data at 255 nm and MS data (FS-IEC of the molecular ions, m/z=301 a.m.u. and m/z=285 a.m.u.), in correlation with retention and demonstrate the use of berries from *Rhamnus* species.

In a visual yellow–orange sample ( $555_{P2}$ ), emodin, rhamnetin and "soluble redwood" (also called "type C", m/z=243) [36, 37] were detected as major dyes and quercetin as minor dye (Fig. 5). As for the other major dyes, the presence of emodin, rhamnetin and "soluble"

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redwood" was put forward by the UV-Vis and MS detectors, while the identification of quercetin was only made by MS data. Emodin (m/z=269 a.m.u.), an anthraquinone dye, may be found in Rhamnus bark, Rheum and Rumex species. Recent studies, performed on Rhamnus frangulae bark and berries, showed that, for the former, mainly emodin together with rhamnetin and quercetin in less amount should be expected [38], which is exactly the case in our study. "Soluble redwood" is a marker for the use of redwood (Caesalpinia species) [17], which could still be identified in redwood dyeing samples when brazilein, the main compound in Caesalpinia species, is too hardly degraded to be identified. This marker compound was recently identified as the benzochromenone urolithin C [39]. Although light fugitive, redwood is a source of red dyes, which would explain the orange hue of sample 555\_ P2. According to the dyes detected, sample 555\_P2 was dyed with a combination of emodin based dye (bark of Rhamnus, Rheum or Rumex species) and redwood (Caesalpinia species).

Fisetin and sulphuretin were detected as major dyes in a visual yellow sample (554\_P5) from a belt decoration (Fig. 6). Their presence is suggested by the UV–Vis spectra and MS data (molecular ions of m/z=285 a.m.u. and m/z=269 a.m.u. in the respective FS-IECs), correlated with retention. According to literature, the combination of the two dyes suggests the use of young fustic (Cotinus coggygria).

#### Shirts

Luteolin was detected as major dye component in three samples (44\_P5, 44\_P8 and 20\_P7) from shirts decorations, in all cases being accompanied by apigenin, which is major dye in 44\_P8 and minor, in 20\_P7. Chrysoeriol was detected as minor compound in one of these samples (44\_P8) which confirm the use of weld (*Reseda luteola* L.). In one other sample (20\_P7), genistein, chrysoeriol, diosmetin and 3-O-methylquercetin were detected (as minor dyes), which indicate that a combination of dyer's broom (*Genista tinctoria* L.), and sawwort (*Serratula tinctoria* L.) was responsible for the colour. For sample 44\_P5, where luteolin was also detected, no other dyes were present, which made impossible the identification of the dyeing source.

## **Red dyes**

#### Belts

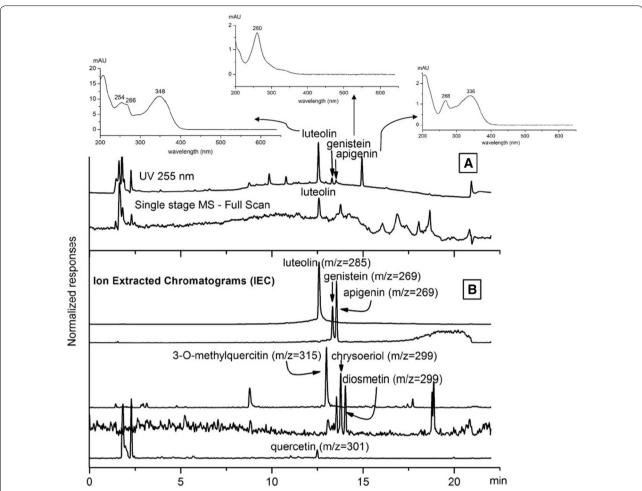
Carminic acid was identified by both UV–Vis and MS detectors in a light red sample (554\_P4) from a belt decoration, the only red sample available from the belts decoration (Fig. 7, left). Its presence was suggested by the UV–Vis spectra correlated with retention and confirmed by the molecular ion m/z=491 a.m.u., in the FS-IEC.

Although considered a minor component in carminic acid containing insects, flavokermesic acid C glycoside, also called dcII [26], was identified in sample 554 P4 by both UV-Vis and MS detectors (m/z=475 a.m.u. in the FS-IEC), which would qualify it as major dye. In the same sample, flavokermesic and kermesic acids were identified as minor components as suggests the presence of their molecular ions, m/z=313 a.m.u. and m/z=329 a.m.u., in the FS-IECs. Carminic acid is the main dye component in Porphyrophora species-originary from the Old World and in Dactylopius coccus (Mexican Cochineal), from the New World. Porphyrophora polonica (Polish carmine scale insect) and Porphyrophora hamelii (Armenian carmine scale insect) are the most well known representatives of their species and have a large geographical distribution, from Switzerland, Germany and Czech Republic to Russia, Kazakhstan and Mongolia [14]. Dactylopius coccus may be found in Mexico, South America and the Canary Islands [14]. Dc II, flavokermesic and kermesic acids co-exist, in various amounts, in all the above-mentioned insect dyes, so that identification of the biological source down to the species level becomes a difficult task. Extremely valuable methods to distinguish between the species, according to the semiquantitative evaluation of the ratio between the three dye components were established by Wouters and Verhecken in 1989 [40, 41] and Serrano et al. in 2015 [42]. Although, in our case, it was not possible to reproduce any of these methods, as they are based on many analysis on standard dyed fibres (fibres dyed in the laboratory according to historical recipes), a simple estimation is possible, even on the limited number of samples analysed in our laboratory. Analysis performed on samples dyed with Porphyrophora species and Dactylopius coccus showed that, for the same amount of sample, the presence of dcII could be clearly observed, with both UV-Vis and MS detectors, for samples dyed with Mexican Cochineal, while it is completely absent in those where Porphyrophora species were used (Fig. 7). Consequently, identification of dcII (flavokermesic acid C glycoside) as major dye, together with carminic acid, and having kermesic and flavokermesic acids as minor dyes, suggest that Dactylopius coccus (Mexican Cochineal) was used for dyeing the light red sample 554\_P4.

## Shirts

Carminic acid was detected as major dye component in all 6 samples (27\_P3, 165\_P3, 44\_P2, 44\_P6, 20\_P1, 20\_P2) which have a visual red, light red or violet colour from the shirts decoration. As already mentioned, it was also detected in the black sample (20\_P3) where Prussian blue was identified (see blue dyes). Except for the light red sample (20\_P2), dc II, kermesic and flavokermesic

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**Fig. 4** UV–Vis spectrum, UV chromatogram (255 nm), total ion current (TIC) chromatogram (resulting after single stage full scan MS detection) and ion extracted chromatograms (m/z = 285, 269, 299, 301, 315 a.m.u.) indicating a combination of dyer's broom (*Genista tinctoria* L.) and sawwort (*Serratula tinctoria* L.) being responsible for the colour in sample 554\_P2 from belt decoration

acids were also detected (see the discussion above). Considering the detection of dc II as major dye as criteria for the attribution of Dactylopius coccus (Mexican Cochineal), the above mentioned would be the most possible dye source for sample 44\_P6. Contrarily, the absence of dc II from the UV-Vis chromatogram, although present in the FS-IEC according to m/z = 475, would indicate the use of Porphyrophora species. Moreover, according to literature and confirmed by analysis performed on standard dyed fibres, kermesic acid is a major dye component in *Porphyrophora polonica* (Polish carmine scale insect) and a minor one in P. hamelii (Armenian carmine scale insect). Consequently, P. hamelii would be responsible for dyeing in samples 27\_P3, 165\_P3, 44\_P2, and 20\_P1. No attribution could be made for sample 20\_P2, where only carminic acid was identified. For all the samples where dyeing was attributed to P. hamelii, as well as in the sample where only carminic acid was detected, ellagic acid was also present, which suggests the use of tannin sources.

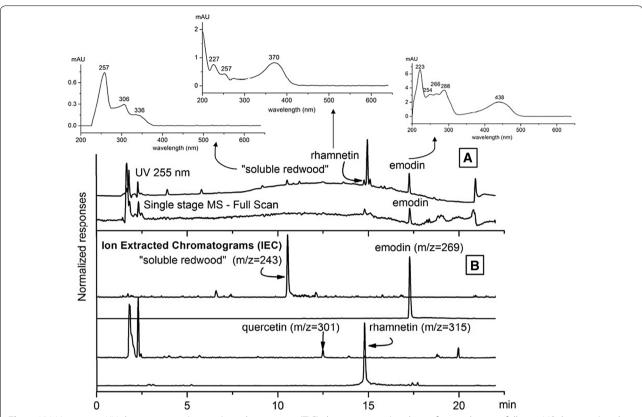
Detection of tin, by XRF analysis, in the black wool sample 20\_P3, where Prussian blue and carminic acid were responsible for dyeing, suggests the use of tin chloride as mordant for the insect dye dyeing. Literature mentions that tin was used as mordant for Mexican Cochineal dyeings from the seventieth century [14], which would indicate that *Dactylopius coccus* (Mexican Cochineal) was used for dyeing in this sample.

## Tannins

#### Belts

No tannins were identified in the samples available from belts decoration.

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**Fig. 5** UV–Vis spectra, UV chromatogram (255 nm), total ion current (TIC) chromatogram (resulting after single stage full scan MS detection) and ion extracted chromatograms (m/z = 243, 269, 301, 315 a.m.u.) illustrating the presence of emodin, rhamnetin and "soluble redwood" (as major dyes) and quercetin (as minor dye) in the analyzed sample (555\_P2)

#### Shirts

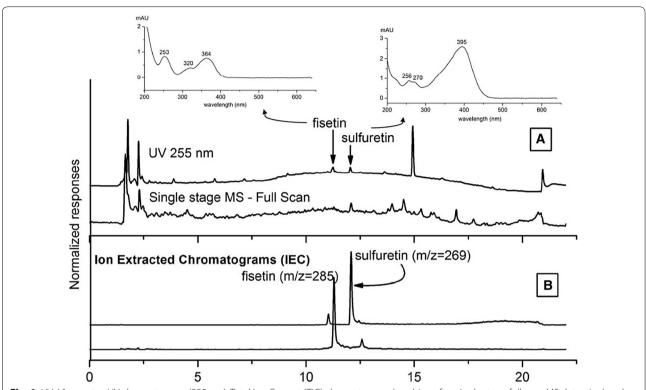
Ellagic acid was detected in 10 silk samples (27 P1, 27\_P2, 27\_P3, 165\_P2, 165\_P3, 165\_P4, 44\_P1, 44\_P2, 20\_P1, 20\_P2) from shirts decoration, which suggests the use of tannin sources. With 3 exceptions (44\_P2, 20\_P1, 20\_P2), its presence was evidenced by the UV-Vis spectra, correlated with retention and confirmed by the molecular ion m/z = 301 a.m.u. in the FS-IECs, which qualifies ellagic acid as major dye. These features correspond to the brown samples, where it was the only component identified, and the red and violet samples where it accompanies Armenian carmine scale insects. In the other cases, mostly characterized as light red, it is only a minor source, identified by MS data. For the cases where no other biological sources were used, tannins should be considered as source of colour (very probable with an iron mordant). Many such dyeing recipes describing the use of tannins with iron mordant to achieve grey, brown or black colours are mentioned in literature [14, 15]. By contrary, detection of ellagic acid in the presence of carminic acid insect dyes suggests that tannins were used for silk weighting [14, 15].

#### Discussion about the dyes and biological sources detected

A large number of biological sources was identified in the belts and shirts decoration, which is well correlated with literature which states that natural dyes were still in use in late nineteenth to early twentieth century, when synthetic dyes were already available [1, 9]. Woad (Isatis tinctoria L.), dyer's broom (Genista tinctoria L.), sawwort (Serratula tinctoria L.), young fustic (Cotinus coggygria Scop.), Rhamnus berries and emodin based dyes (Rhamnus, Rheum, Rumex sp.) are local sources while carminic acid containing insects (Dactylopius coccus Costa and Porphyrophora sp.) and Caesalpinia species are a consequence of trade. Weld (Reseda luteola L.) is a rare plant on the Romanian territory, in contrast to its relative, Reseda lutea L., more frequent but poorer in dyes [14]. Consequently, the detection of weld in belts and shirts decoration should be also considered a result of commerce.

Most of the biological sources identified in the present study, with reference to Transylvania (western part of Romania), were also detected in previous work on ethnographical textiles from Moldavia and Wallachia (Southern and Eastern parts of Romania, see Fig. 1),

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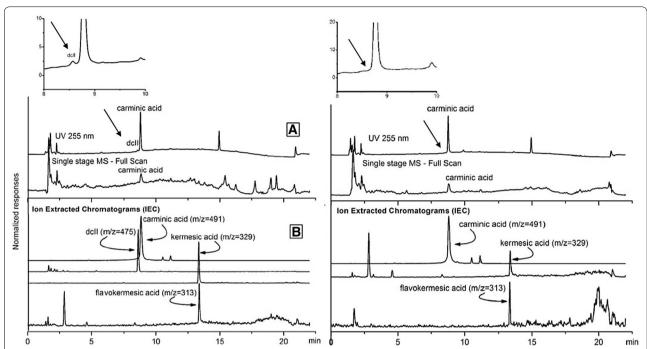
**Fig. 6** UV–Vis spectra, UV chromatogram (255 nm), Total Ion Current (TIC) chromatogram (resulting after single stage full scan MS detection) and ion extracted chromatograms (m/z=285, 269 a.m.u.) illustrating the presence of fisetin and sulfuretin as major dyes in a visual yellow sample (554\_P5) from a belt decoration

from the same period [7]. Although difficult to be translated in a modern language, valuable information about the art of dyeing, at the end of the nineteenth century and beginning of the twentieth century, is offered by a collection of recipes published by the Romanian Academy in 1914, with reference to Moldavia and Wallachia [9]. Written with the intention to keep alive the memory of the declining art by putting together the existing recipes, the book allows us understand which were the natural and synthetic dye sources and how frequently they were used. For example, dyer's broom ("drobiţă", "droghiță"), a local spontaneous plant, appears in more than 60% of the recipes where yellow dyes are concerned, which qualifies it as the most frequent source of yellow. Other recipes describe the use of various local sources - including the ones identified in the present study dedicated to dyes in belts and shirts decoration in Transylvania: sawwort ("gălbinare", "șoldeală"), young fustic ("scumpie"), Rhamnus berries ("verigariu"), emodin based dyes ("cruşîn"), woad ("drobşor"). A careful reading of the recipes which describe "cârmâz" in association with tin reveals the use of Mexican Cochineal (Dactylopius coccus) as imported natural source. Dyeing with Prussian blue ("lulachiu") is also described, although the respective recipes only refer to dyeing in blue and green. The only dyes detected in the present study for which no reference was found in the book are weld (*Reseda luteola* L.), redwood type (*Caesalpinia* species) and indigo carmine [9].

#### **Conclusions**

Several biological sources were identified in belts and shirts decoration from Transylvania, dating from late nineteenth to early twentieth century. Most of them, such as dyer's broom (Genista tinctoria L.), sawwort (Serratula tinctoria L.), young fustic (Cotinus coggygria Scop.), Rhamnus berries, emodin based dyes (Rhamnus, Rheum, Rumex sp.) should be considered local sources while others were imported. The latter include redwood type (Caesalpinia sp.) and carminic acid containing insects (Dactylopius coccus Costa and Porphyrophora sp.). Weld (Reseda luteola L.), also detected in two cases, is considered a rare plant in Romania, which indicates its presence to be a result of commerce. Woad (Isatis tinctoria L.) is mentioned as local but it is not possible to establish, based on the existing methods, if it or imported indigo was used in hemp belts or wool and silk belts and shirts decoration. Early synthetic

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**Fig. 7** UV–Vis spectrum, UV chromatogram (255 nm), total ion current (TIC) chromatogram (resulting after single stage full scan MS detection) and ion extracted chromatograms (m/z=313, 329, 475, 491 a.m.u.) indicating the presence of dcll (flavokermesic acid C glycoside) and carminic acid (as major dyes), together with kermesic and flavokermesic acids (as minor dyes), suggesting *Dactylopius coccus* (Mexican Cochineal) as dyeing agent for the light red sample 554\_P4

dyes, such as indigo carmine and Prussian blue were also detected. The interest to obtain a large variety of hues in belts decoration may be deduced from the number of flavonoid dyeing sources detected, in contrast to the red and brown dominant colours in shirts decoration, for which almost only carminic acid based insects and tannin sources were used.

As compared with earlier results on ethnographical textiles in Moldavia and Wallachia from the same period [7, 8], it could be stated that the same natural dye sources were detected in textiles from the three provinces. However, at a closer look, the large palette of colours found in only 10 samples from Transylvanian belts decoration is obviously remarkable.

Although the number of synthetic dyes detected is too low to draw any conclusion, it is worth to underline that Prussian blue was only detected in textiles from Transylvania. All the results obtained are in perfect agreement with literature, if we consider that almost all the natural and synthetic dyes detected are frequently mentioned in a collection of recipes published by the Romanian Academy, in 1914 [9]. The richness in colors in belts, the use of insect dyes in shirts decoration and the large amount of cotton in shirts are illustrative for the owners status and for the interest of ASTRA (Association) in 1905, to collect highest level objects for their museum.

The analytical protocol developed, based on the combined use of the UV–Vis and mass spectrometric detectors to associate the information and distinguish between major and minor dyes, facilitates a clear attribution of the dyes and biological source/sources used. Other techniques, such as X-ray spectroscopy and FTIR-ATR were successfully used to identify inorganic dyes, which may not be detected by LC–DAD–MS, as was the case of Prussian blue.

Scientific certification of the objects value together with confirmation of dyeing sources in a collection of recipes published 100 years ago provide a better valorisation of the Romanian traditional costume as witness of the rural society at the end of the nineteenth to beginning of the twentieth century and emphasizes, once more, the usefulness of chemistry in cultural heritage dedicated applications.

#### **Abbreviations**

a.m.u.: atomic mass unit; ATR: attenuated total reflectance; DAD: diode array detection; FS: full scan mode; FTIR: Fourier transform infra red spectrometry; HPLC: high performance liquid chromatography; UHPLC: ultra high pressure liquid chromatography; IEC: ion extracted chromatogram; m/z: mass to charge ratio; MS: mass spectrometry; MS/MS: tandem MS spectrometry; XRF: X ray fluorescence spectrometry.

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#### Authors' contributions

IP and IT designed the research; IT performed sampling and provided data about the objects; MV provided the necessary instrumentation and assisted IP during sample preparation; FA and AM set up, together with IP, the LC–DAD–MS analytical protocol and provided all the support for LC and MS data acquisition and procession; EN assisted the team with information about local and imported vegetal dyeing sources. All authors read and approved the final manuscript.

#### Author details

<sup>1</sup> National Museum of Romanian History (MNIR), Calea Victoriei 12, 030026 Bucharest, Romania. <sup>2</sup> The ASTRA National Museum Complex, Piata Mica 11, 550182 Sibiu, Romania. <sup>3</sup> Agilrom Scientific SRL, Bucharest, Sos. Pipera-Tunari 1H, 077190 Voluntari, Ilfov, Romania. <sup>4</sup> "Horia Hulubei" National Research Institute for Physics and Nuclear Engineering, IRASM, Str. Reactorului 30, Magurele, Romania. <sup>5</sup> University of Bucharest, "Dimitrie Brândză" Botanical Garden, Sos. Cotroceni 32, 060114 Bucharest, Romania. <sup>6</sup> Department of Analytical Chemistry, University of Bucharest, Faculty of Chemistry, Sos. Panduri 90, 050663 Bucharest, Romania.

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The authors declare that they have no competing interests.

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